

Synthesis, Spectral Characterization and Anticancer Study of Bio-Complexes Pd from Methyl Imidazole

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Abstract

The prepared bio- ligand derived from chalcone -azo (E)-1-(4-(E)-4-methyl-derivative of di-nitrogen atoms-cycle. In view of the effect of the compensating groups on the effectiveness and uses of the five-cyclic heterogeneous compounds, in this research new substitutes for the aromatic azo and chalcon compounds were prepared from different raw materials and compounds that could have different biological effects and effects on the cell of the organism were prepared and their physical and spectral properties were studied. The importance of studying the spectral characterization of a number of prepared compounds containing different groups of compensators, which may have an impact in the inorganic chemistry field or the industrial field., In addition to anti-cancer study against cancer cell.

Key Words: chalcone , azo, 4- methyl imidazole, inorganic.

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INTRODUCTION

It could be noticed that, a better understanding of stability and intense color within the visible area of complexes which derived from azo imidazole due to unpaired electrons (Abate, 2020 .; Abd, 2022) of azo-imine functional groups that it coordinated with empty orbitals of metal ions (Afaq , 2019 ., Altaie, 2022). All chalcone compounds consider as catalysts, antimicrobials , colorants, corrosion (Al-Daffay, & Alhamdani, 2022 .; Alina, 2021), inhibitors and in optical physics , printer and electronic fields and food industry, therefore the complexes show high inhibition activity (Bouhdada, 2022 .; Dian, 2022.; Djmal, 2022) against breast cancer line (MCF-7) .In addition, complexes derived from chalcones (Cleasen-schmedit

condensation) of acetophenone derivatives and benzaldehyde derivatives will be given (α,β – unsaturated carbonyl compounds) have an important role in coordination of metal ions and as anticancer. Palladium complexes with azo ligands have been investigated as antitumor . The majority of Palladium complexes were found to be more active than their free ligands (Göknıl, et al 2022 .; Hussein, & Khalid , 2021 .; Hussein, & Abbas, 2017).Derivative of imidazole and chalcone and attempted preparation and identification some of metal ions which coordinated with it (Najiyah, et al 2022 .; Mahmood, & Jawad 2021) .

METHODOLOGY

Constituents and quantities

All substance materials and diluents which have been used in this search in high purity. In addition, number of measurements have been done such as technical spectral instruments.

Combination of (LMN) chalet

This chalet was combined by dual ladders: Firstly, (Cleasen-schmedit condensation) [16]by solving (4.05g, 0.02 mole) carbonyl – amine dreivative with (60ml) basic solution with emotive and cooling in the slush soak at (some minutes.) . Then, an basic elucidation of 4-Nitrobenzaldehyde (4.5g, 0.03 mole) was added slowly with exciting at (7hrs.) at apartment heat. The thickened solution was formed , which it was kept overnight . Following , diazotate of selected compounds bestowing to readings it (Najiyah, et al 2022 .; Mahmood, & Jawad 2021) .,figure (1)

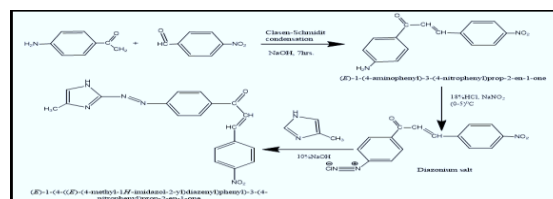


Fig.1. Synthesis (LMN) ligand.

Preparation of Complexes

All solid Chaletes-ion were prepared by accumulation acetone clarification of ligand (LMN) to elucidations ions (II) with acetonitrile elucidation of each ion (II) in mole share (L:M) as (2:1) through thrilling and thermo-method about (some minutes.) , the dyed participations were collected , desiccated and crystallized by burning solvent.

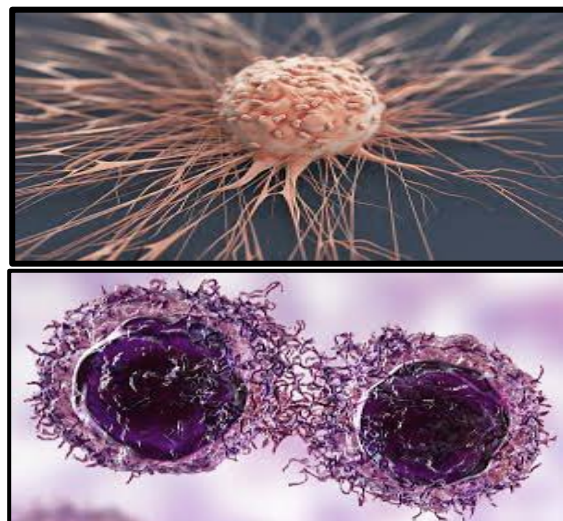
Tab.1. Amount of some properties.

Chalets	M wt	Dye d	Mel ting	Yi el d %	(Found) Calc.%			
					% C	% H	% N	% M
LMN= C ₁₉ H ₁₅ N ₅ O ₃	36 1.3 5	Dar k- ora nge	102 - 105	80	63	4.	19	
					.0	1	.3	
					9	5	7	---
					-	-	-	--
[Co(C ₁₉ H ₁₅ N ₅ O ₃) ₂ Cl ₂]	85 2.6 3	Dar k red dish - Bro wn	204 - 208	79	53	3.	16	6.9
					.4	5	.4	1
					8	1	1	-
					-	-	-	-
[Ni(C ₁₉ H ₁₅ N ₅ O ₃) ₂ Cl ₂]	85 2.3 9	Red dish bro wn	220 - 224	77	53	3.	16	6.8
					.4	5	.4	8
					9	1	2	-
					-	-	-	-
[Cu(C ₁₉ H ₁₅ N ₅ O ₃) ₂ Cl ₂]	85 7.2 4	Dar k Bro wn	198 - 200	89	53	3.	16	7.4
					.1	4	.3	1
					9	9	3	-
					-	-	-	-
[Zn(C ₁₉ H ₁₅ N ₅ O ₃) ₂ Cl ₂]	85 9.0 9	Red dish Bro wn	212 - 218	76	53	3.	16	7.6
					.0	4	.2	1
					7	9	9	-
					-	-	-	-
[Cd(C ₁₉ H ₁₅ N ₅ O ₃) ₂ Cl ₂]	90 6.1 1	Lig ht red dish Bro wn	198 - 202	78	50	3.	15	12.
					.3	3	.4	4
					2	1	5	-
					-	-	-	-
[Hg(C ₁₉ H ₁₅ N ₅ O ₃) ₂ Cl ₂]	99 4.2 9	Dar k bro wn	185 - 188	88	45	3.	14	---
					.8	0	.0	---
					6	1	8	-
					-	-	-	-
					45	2.	13	-
					.7	9	.9	-
					9	8	8	-
					-	-	-	-

[Pd(C ₁₉ H ₁₅ N ₅ O ₃) ₂ Cl ₂]	90 0.1 2	Red dish bro wn	215 - 218	74	50	3.	15	11.
					.6	3	.5	82
					5	3	5	-
					-	-	-	-
					50	3.	15	11.
					.4	2	.4	78
					9	9	9	-
					-	-	-	-

Type of Cancer Cell

The type of cancer cell in natural :



RESULTS AND DISCUSSION

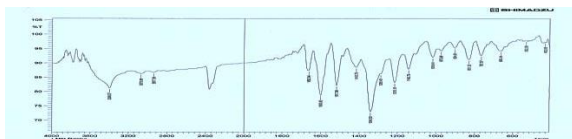
FTIR Varieties of (LMN) chalet and its relations

The FTIR spectrum of free l chalet appeared strong band at (1600)cm⁻¹ referred to carbonyl group of chalcone , this band did not grieve any changes in varieties of relation because of incoordination. However, both bands (N=N) and (C=N) have publicized plentiful variations in shapes, frequencies in addition to intensities in the spectra of all chalets due to the management with chalet . In addition, there are new bands noticed in all complexes due to (M-N) group.

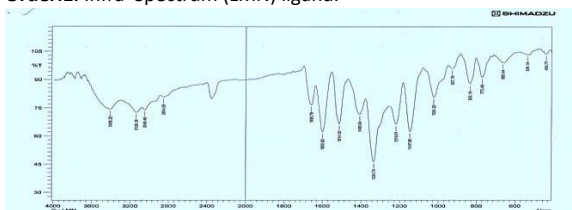
Tab.2. Amount of some properties.

M-N	v(C=O) Chalcone	N=N	C=N)v	v (N- H)	Compound
.....	1600	1415	1517	3398	LMN
440	1600	1409	1510	3394	[Co(LMN) ₂ Cl ₂]
443	1600	1411	1512	3423	[Ni(LMN) ₂ Cl ₂]
439	1600	1408	1514	3408	[Cu(LMN) ₂ Cl ₂]
432	1600	1413	1514	3392	[Zn(LMN) ₂ Cl ₂]

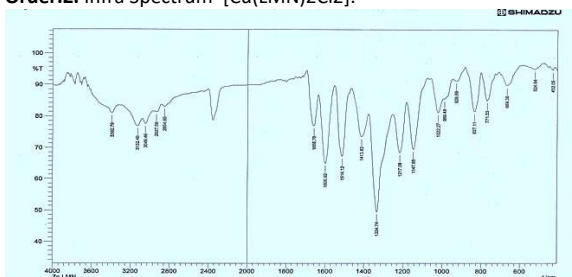
453	1600	1413	1516	3402	[Cd(LMN) ₂]Cl ₂
449	1600	1411	1510	3390	[Hg(LMN) ₂]Cl ₂
432	1600	1408	1519	3444	[Pd(LMN) ₂]Cl ₂



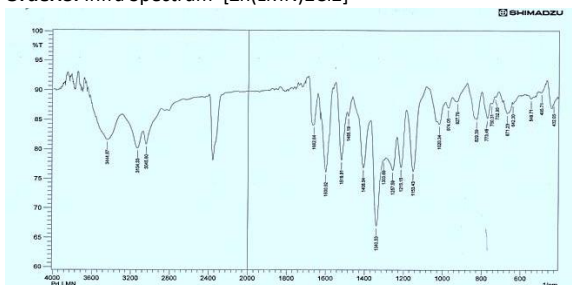
Order.1. Infra Spectrum (LMN) ligand.



Order.2. Infra Spectrum [Cu(LMN)2Cl2].



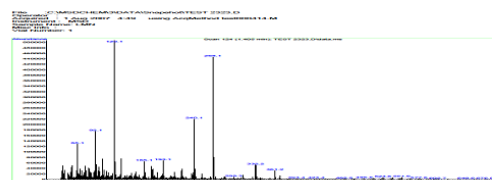
Order.3. Infra Spectrum [Zn(LMN)2Cl2]



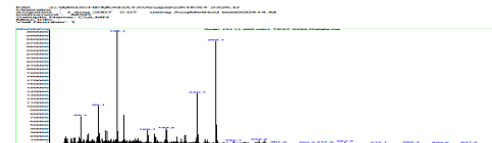
Order.4. Infra Spectrum [Pd(LMN)2]Cl2

Mass gamut

Mass gamut of new chalet (LMN) in addition to Cu(II) chalet remain publicized in shape.(5) besides (6) However, the suggestion fragmentation of ligand is presented in scheme(2), which is appeared base beak at (M/Z*) (361.2) for mother molecule and other mass spectral fragmentations, which are agreement the formula.



Order.5: Mass spectra (LMN) chalet



Order.6: Mass spectra of Cu(II) complex

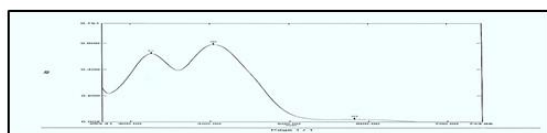
Electronic Spectra

This study compared the gamut of permitted chalet per the varieties of its multiplexes, all compounds were dissolved in DMSO at (1x10⁻⁴)M, the result show red shift in electronic spectral of all complexes, which is referred to coordination with ligand, the data of all compounds is consisted in table (4).

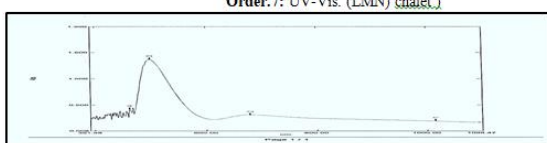
Tab.3. Amount of some properties.

Chalet	λ_{max}	Absorption (cm ⁻¹)	Conversion	μ_{eff} (B.M)	shape	Hybridization	Conduct S.mol ⁻¹ .cm ²
	(nm)						
LMN	4044	24752	n → π*				
	3266	30674	π → π*	---	----	----	----
	2445	40816	π → π*				
[Co(LMN) ₂]Cl ₂	679	14727	⁴ T _{1g(F)} → ⁴ A _{2g(F)}				
	463	21598	M.L.CT	4.7	Octahedral	Sp ³ d ²	38.3
	346	28901	I.L.C.T.				
[Ni(LMN) ₂]Cl ₂	508	19230	M.L.CT				
	342	29239	I.L.C.T.	2.64	same	Sp ³ d ²	43.4
	290	34482	I.L.C.T.				
[Cu(LMN) ₂]Cl ₂	670	14925	² B _{1g} → ² E _g				
	522	19157	M.L.C.T.	1.81	same	Sp ³ d ²	8.5

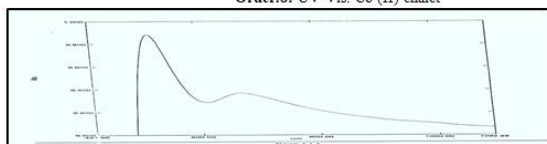
[Zn(L MN) ₂ C l ₂]	4 9 6	2016 1	M.L C.T.	Di a	Sam e	Sp ³ d ²	25
[Cd(L MN) ₂ Cl ₂]	5 0 2	1992 0	M.L C.T.	Di a	Sam e	Sp ³ d ²	16. 9
[Hg(L MN) ₂ C l ₂]	5 0 9	1964 6	M.L C.T.	Di a	Sam e	Sp ³ d ²	22. 5
[Pd(L MN) ₂ Cl ₂]	5 2 8	1893 9	¹ A _{1g} → ¹ B _{1g}	Di a	Squa re plan er	dsp ²	72. 16



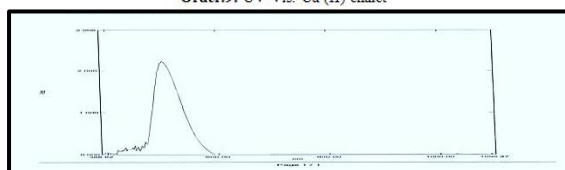
Order.7: UV-Vis. (LMN) chalet.)



Order.8: UV-Vis. Co (II) chalet



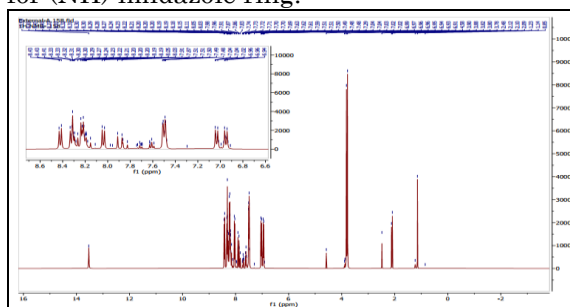
Order.9: UV-Vis. Cu (II) chalet



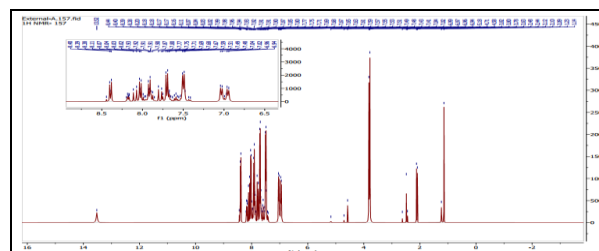
Order.10. UV-Vis. Cd (II) chalet

H-NMR Varieties

¹H-NMR Varieties of ligand (LMN) and Hg(II)chalet characterized in DMSO-d₆ solvent and TMS reference both spectra appeared a signals at (6.9,6.7)and (7.1,7.3)ppm owed to (CH=CH-CO) of chalcone derived [17], and signal at (2.3,2.4)ppm credited to (CH₃) cluster in imidazole sphere [15] . While, multiples protons around (7-8)ppm denoted to aromatic rings and signal at (13.8, 13.7)ppm for (NH) imidazole ring.



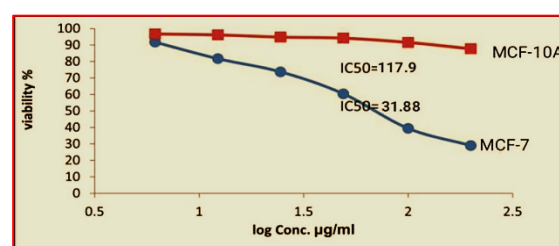
Order.11. Resonance (LMN)ligand



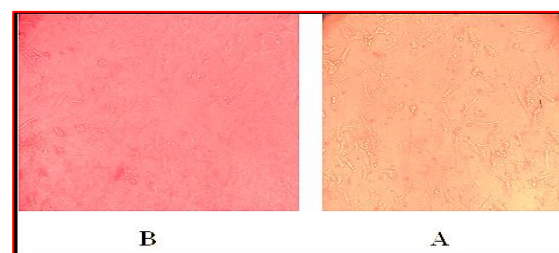
Order.12. Resonance Hg(II)complex

Cytotoxic Activity of Complex

The result of this study investigated that, IC₅₀= (31.88) breast tumor line(MCF-7) . However, IC₅₀= (117.9) for well cells line(MCF-10A) ,it indicated Pd(II) chalet could be charity as suggested medicine of this kind of cancer.



Order.3. activity-Tumor of Pd(II) chalet



Order.4. (A)(MCF-7) and(B) (MCF-10A) Under inverted microscope.

CONCLUSIONS

This search involved synthesis of different chalet chalcone besides imidazole and prepared its relations presented an Octahedral except Pd(II) complex which took square planer . Additionally, this search appeared sharp bands that indicated to new synthesized ligands. Also this bio-compound gave good results against of cancer cell.

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AUTHOR CONTRIBUTIONS

The author carried out : Conceptualization, Methodology, Investigation, Software.

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There is no backing.

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